

Fire Extinguisher. Task:

Use the sheet attached to make your own fire extinguisher. Then select your own type of written work from the list below.

How could you adapt this experiment?

- Use different quantities of vinegar or bicarbonate of soda.
- Time how long it takes for the flame to go out under different conditions.
- What other variables could you change?

DO NOT ATTEMPT THIS WITHOUT AN ADULT PRESENT!

Task Selection

- 1) Write an explanation about why this would be a better way of putting out fires than water?
- Write-up your experiment using the structure you've used before.
- 3) Write a persuasive leaflet to influence people to fit and check smoke alarms.

FIRE EXTINGUISHER

SCIENCE CHALLENGE

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Designed by Liam, Design engineer at Dyson

The brief

Create your own invisible fire extinguisher.

The method

- 1. With the help of an adult, light the candle.
- Mix a little bicarbonate of soda and vinegar together in the jar to make a frothing mixture.
- Tip the jar over the candle so only the gas from the reaction comes out. Be careful not to tip the mixture out.
- The flame will be extinguished.



Materials

Matches (with adult supervision) Bicarbonate of soda Vinegar

Candle

A jam jar



Design icons

Dry chemical extinguishers are filled with powder, which is usually sodium bicarbonate or baking soda. When released over the fire, the powder decomposes at 70°C releasing CO₂, smothering the fire.



How does it work?

The mixture of bicarbonate of sada and vinegar creates carbon dioxide.

CO₂ is heaver than air so it sits at the bottom of the glass. When you tip up the glass, the CO₂ comes out and suffocates the candle.